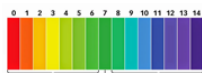


Chemistry at WOOTTON ACADEMY TRUST

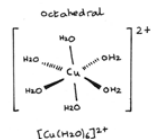


$$K_c = \frac{[NH_3(g)]^2}{[N_2(g)][H_2(g)]^3}$$

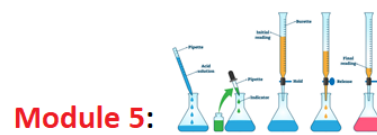
Module 5:
Equilibrium



Module 5: Acids, Bases
and Buffer Solutions

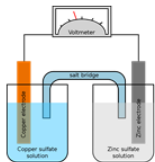


Module 5:
Transition Elements



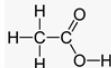
Module 5:
Redox + Redox Titrations

Module 5:
Electrode
Potentials



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Module 6: Carbonyls
+ Carboxylic Acids



Module 6:
Amines, Amino
Acids + Polymers

Module 6:
Organic
Synthesis

Module 6:
Chromatography +
Spectroscopy

Module 5: $\Delta G = \Delta H - T\Delta S$
Enthalpy + Entropy

Module 4: Synthesis
+ Spectroscopy

Module 4: Alcohols
+ Haloalkanes

Module 4:
Alkanes + Alkenes

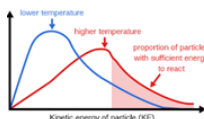
Module 4: Basic Concepts
of Organic Chemistry

Module 2: 6.022×10^{23}
Amount of
Substance,
Acids + Redox

Module 6:
Aromatic
Chemistry



Module 5:
Rates of Reactions



Module 3: Enthalpy,
Reaction Rates +
Equilibrium

Module 3:
Periodicity +
Reactivity Trends

PERIODIC TABLE OF ELEMENTS

Module 2: Atoms, Ions,
Electrons, Bonding and
Intermolecular Forces

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